

# Physics

**Area:** Area is the amount of surface enclosed within the boundary lines.

**Dispersion:** This is the splitting up of white light into separate colours. It can be done by passing white light through a prism.

**Energy:** Energy is the ability to do work.

**Equilibrium:** An object that is balanced is said to be in equilibrium.

**Force, F:** A force is that which causes a change in the velocity of an object.

**Unit:** Newton, N

**Formula:** Force = Mass x Acceleration ( $F = ma$ )

**Freezing:** This is the changing of a liquid to a solid state.

**Frequency, f:** This is the number of waves that pass a particular point in one second.

**Friction:** This is a force which opposes motion between two objects in contact.

**Fuse:** A fuse is a safety device in an electric circuit. If the current gets too high the wire in the fuse melts which breaks the circuit switching off the current.

**Galaxy:** A large group of stars held together by its own gravity.

**Heat:** Heat is a form of energy.

**Unit:** Joules, J

**Insulator:** This is a substance, which does not allow heat to flow through easily.

**Latent heat:** This is the heat absorbed or released when a substance changes state without changing temperature.

**Law of conservation of energy:** Energy cannot be created or destroyed but can be converted from one form to another.

**Law of the lever:** When a lever is balanced the sum of the clockwise moments is equal to the sum of the anti clockwise moments.

**Lever:** A lever is a rigid body, which is free to turn about a fixed point called the fulcrum.

**Light:** Light is a form of energy.

**Loudness:** The loudness of a sound depends on the amplitude

**Lubricant:** A lubricant is a substance capable of reducing friction.

**Luminous:** A luminous object is an object that gives out light.

**Lunar eclipse:** This happens when the earth passes between the sun and the moon.

**Magnetic field:** A space around a magnet in which the magnetism can be detected.

**Mass, m:** The mass of an object is the quantity of matter in it.

**Melting:** This is the changing of a solid to a liquid state.

**Moment:** This is a measure of the turning effect of a force.

**Formula:**

Moment of a force = Force x Perpendicular distance from the fulcrum.

**Newton's third law of motion:** For every action there is an equal but opposite reaction.

**Ohm's law:** At constant temperature the voltage across a conductor is proportional to the current flowing through it.

**Formula:** Voltage = Current x Resistance ( $V = IR$ )

**Pitch:** The pitch of a sound is how high or low it is. It depends on the frequency of the wave.

**Potential difference:** Potential difference is also called voltage. It is the force, which moves the electrons around the circuit.

**Unit:** Volt (V)

**Power:** This is the rate at which energy is converted from one form to another.

**Unit:** Watts (W)

**Formula:** Power = Voltage x Current ( $P = VI$ )

**Pressure:** Pressure is force per unit area.

**Formula:** Pressure =  $\frac{\text{Force}}{\text{Area}}$   $\left( P = \frac{F}{A} \right)$

**Unit:**  $\text{N/m}^2$  or Pascal (Pa)

**Primary colours:** The primary colours are red, green and blue. When the three of these colours are combined it results in white.

Red + Green + Blue = White

**Radiation:** This is the transfer of heat by means of invisible rays, which travel outwards from the hot object, without needing a medium.

**Rectifier:** This is used to convert alternating current to direct current.

**Reflection:** The reflection of light is the bouncing back of light from a surface.

**Refraction:** The refraction of light is the bending of light as it passes from one medium to another.

**Resistance, R:** The opposition of a conductor to current is called its resistance. A good conductor has a low resistance and a bad conductor has a high resistance.

**Secondary colours:** A secondary colour is formed when two primary colours are mixed. The three secondary colours are yellow, magenta and cyan.

**Red + Green = Yellow**

**Red + Blue = Magenta**

**Blue + Green = Cyan**

**Solar eclipse:** This happens when the moon passes between the sun and the earth.

**Sound:** Sound is a form of energy.

**Speed, v:** Speed is the distance travelled per unit time.

**Formula:**  $\text{Speed} = \frac{\text{Distance}}{\text{Time}} \left( v = \frac{s}{t} \right)$

**Unit:** m/s

**Stable equilibrium:** A body is in stable equilibrium if when slightly moved its centre of gravity rises.

**Sublimation:** This is the changing of a solid directly to a gas. (Iodine is an example of a substance that sublimates).

**Temperature:** This is a measure of how hot an object is.

**Unit:** degrees Celsius ( $^{\circ}\text{C}$ )

**Unstable equilibrium:** A body is in unstable equilibrium if when slightly moved its centre of gravity falls.

**Velocity:** Speed in a given direction.

**Units:** m/s

**Volume:** The volume of an object is the amount of space it takes up.

**Wave:** A wave is a means of transferring energy from one point to another.

**Formula:** Velocity = Frequency x Wavelength ( $v = f \times \lambda$ )

**Wavelength:** The wavelength of a wave is the distance between any two successive crests.

**Weight:** Weight is the force of gravity on an object.

**Formula:** Weight = Mass x Acceleration due to gravity

# Chemistry

**Acid rain:** Rainwater with a pH of less than 5.7 is acid rain. It is caused by the gases  $\text{NO}_2$  (from car exhaust fumes) and  $\text{SO}_2$  (from the burning of fossil fuels) dissolving in rain. Acid rain kills fish, kills trees, and destroys buildings and lakes.

**Acid:** An acid is a proton donor. It turns litmus red.

**Activity Series:** The activity series is a list of metals in order of decreasing reactivity.

**Alkali metals:** These are the elements in group one in the periodic table.

**Alkaline earth metals:** These are the elements in group two in the periodic table.

**Alloy:** An alloy is a mixture of metals. Bronze is an example of an alloy it is formed from copper and tin.

**Atom:** An atom is the smallest part of an element, which can exist.

**Atomic number:** The atomic number of an atom is the number of protons in the nucleus of the atom.

**Base:** A base is a proton acceptor. It turns litmus blue.

**Capillarity:** This is the rising of liquids up a narrow tube.

**Chemical change:** A chemical change is one in which there is a new substance formed.

**Cobalt chloride paper:** This paper is used to test for water. If water is present it changes colour from blue to pink.

**Combustion:** Combustion is also called burning. This is the combining of a substance with oxygen.

**Compound:** A compound is a substance made up of two or more elements chemically combined.

**Corrosion:** Corrosion is an undesired process where a metal is converted to one of its compounds, e.g. rusting.

**Covalent bond:** A covalent bond is a force of attraction between two atoms as a result of their sharing of electrons.

**Distillation:** The vaporisation of a liquid by heating and then the condensation of the vapour by cooling.

**Dry cell:** This is a battery in which the electrolyte is in the form of a paste.  
**Ductile:** Metals are ductile. This means they can be pulled out to form wires.

**Electrode:** An electrode is a conductor, which dips into an electrolyte and allows the electrons to flow to and from the electrolyte.

**Electrolysis:** This is the production of a chemical change using electricity. Electrolysis can be used to split up water into hydrogen and oxygen.

**Electrolyte:** An electrolyte is a substance which when dissolved in water conducts electricity.

**Electroplating:** This is where a metal is covered with a layer of another metal using electricity.

**Element:** An element is a substance, which cannot be split up into simpler substances by chemical means.

**Endothermic reaction:** An endothermic reaction is a reaction that takes in heat, e.g. adding water to ammonium chloride.

**Exothermic reaction:** An exothermic reaction is a reaction that gives out heat, e.g. burning of coal.

**Fossil fuels:** Fuels that were formed from the remains of plants and animals that lived millions of years ago.

**Fuel:** A fuel is any substance that burns in oxygen to produce heat.

**Halogens:** These are the elements in group seven in the periodic table.

**Hard water:** This is water that finds it difficult to form lather with soap.

**Immiscible liquids:** These are liquids that do not mix to form a solution, e.g. oil and water.

**Indicator:** An indicator is a substance, which shows by means of a colour change if a substance is acidic or basic.

**Ion Exchange:** This is a method of removing hardness from water. It replaces the positive ions that cause the hardness with  $H^+$  ions.

**Ion:** An ion is a charged atom or group of atoms, e.g.  $Na^+$ .

**Ionic bond:** An ionic bond is a force of attraction that occurs between oppositely charged ions in a compound. It results from a transfer of electrons.

**Joule:** This is the unit of energy and work.

**Malleable:** Metals are malleable. This means they can be hammered into sheets.

**Mass number:** The mass number of an atom is the number of protons and neutrons in the nucleus of the atom.

**Matter:** Matter is anything which occupies space and has mass.

**Miscible liquids:** These are liquids that mix to form a solution, e.g. alcohol and water.

**Mixture:** A mixture consists of two or more different substances mingled together but not chemically combined.

**Molecule:** A molecule consists of two or more atoms chemically combined.

**Neutralisation:** This is the reaction between an acid and a base to give salt and water.

**Octet rule:** During bonding atoms tend to reach an electron arrangement with eight electrons in the outermost shell.

**Oxidation:** Oxidation is the addition of oxygen or the losing of electrons.

**pH scale:** This is a scale from 0 to 14.

If the pH of a solution is 7 it is neutral; if the pH of a solution is less than 7 it is acidic; if the pH of a solution is greater than 7 it is basic.

**Permanent hardness:** This is hardness in water that cannot be removed by boiling. It is caused by calcium sulphate.

**Physical change:** A physical change is one in which there is no new substance formed.

**Products:** These are the chemicals that are produced in a chemical reaction.

**Reactants:** These are the chemicals that react together in a chemical reaction.

**Reduction:** Reduction is the removal of oxygen or the gaining of electrons.

**Salt:** A salt is formed when the hydrogen of an acid is replaced by a metal.

**Saturated Solution:** A solution, which contains as much solute as it can hold at that temperature.

**Solution:** A solution is a mixture of a solute (usually a solid) and a solvent (usually a liquid).

**Suspension:** A suspension is a mixture of a liquid and a finely divided insoluble solid.

**Temporary hardness:** This is hardness in water that can be removed by boiling. It is caused by calcium hydrogencarbonate.

**Titration:** This is the process of adding one solution from a burette, to a measured amount of another solution to find out exactly how much of each is required to react.

**Valency:** The valency of an element is the number of electrons an atom of the element wants to gain, lose or share so as to have a full outer shell.

# Biology

**Absorption:** This is the movement of food into the bloodstream.

**Alkaline pyrogallol:** This is used to absorb oxygen.

**Amylase:** This is an enzyme. It is found in saliva. It breaks starch down into maltose.

**Antagonistic muscles:** A pair of skeletal muscles that work together. When one contracts the other relaxes, e.g. the biceps and triceps.

**Asexual reproduction:** Reproduction that does not involve gametes.

**Assimilation:** This is the using of the food by the cells of the body after absorption.

**Benedict's solution:** This is used to test for a reducing sugar e.g. glucose. If a reducing sugar is present it turns brick red after being heated in a boiling water bath.

**Breathing:** This is a physical process of taking in oxygen and breathing back out carbon dioxide.

**Carnivore:** An animal that only eats other animals.

**Carpel:** The female part of the flowering plant. It is made up of the stigma, style and ovary.

**Catalyst:** A chemical that speeds up or slows down chemical reactions.

**Cell wall:** Structure found outside the cell membrane in plant cells. Cell walls are absent in animal cells.

**Chlorophyll:** The green pigment found in the chloroplasts of plant cells. It is used in photosynthesis.

**Competition:** This is the struggle between organisms to gain a sufficient supply of a scarce resource e.g. Grasses and dandelions compete for water.

**Conservation:** This is the wise use of the environment:

**Digestion:** This is the breaking down of food into small soluble pieces.

**Dispersal:** The dispersal of seeds is the scattering of seeds. The advantage of dispersal is that it helps reduce competition.

**Egestion:** The getting rid of unused, undigested and unabsorbed food material

**Endocrine glands:** A ductless gland that releases hormones directly into the bloodstream, e.g. the pancreas (it releases insulin which controls blood sugar level).

**Excretion:** This is the getting rid of waste products from chemical reactions in the body.

**Food chain:** A food chain is a feeding relationship between organisms through which energy is transferred.

**Food web:** A food web is a number of interconnected food chains.

**Gamete:** A gamete is a sex cell. The male gamete is the sperm and the female gamete is the egg.

**Genetics:** This is the study of inheritance.

**Geotropism:** The growth of a plant in response to gravity.

**Germination:** Germination is the growth of a seed into a new plant. The requirements are warmth, moisture and oxygen.

**Habitat:** The place where a plant or animal lives is called its habitat.

**Haemoglobin:** The red pigment in red blood corpuscles. It is involved in transporting oxygen.

**Herbivore:** An animal that eats only plants.

**Hormone:** A chemical substance that is released by an endocrine gland.

**Humus:** The organic material of soil. It is formed from decomposing plants and animals.

**Implantation:** This is when the embryo attaches itself to the womb wall.

**Ingestion:** This is the taking in of food into the mouth.

**Iodine solution:** This is used to test for starch. If starch is present it turns blue-black.

**Iris:** The iris controls the amount of light entering the eye.

**Joint:** This is where two or more bones meet.

**Leaching:** The washing of minerals out of the soil.

**Ligaments:** Fibres that connect bone to bone.

**Lime water:** This is used to test for the presence of carbon dioxide. If carbon dioxide is present the lime water turns milky.

**Motor nerve:** A nerve that carries messages away from the brain and spinal cord.

**Nutrient agar:** This is used as a food supply for bacteria and fungi in the lab.

**Omnivore:** An animal that eats plants and animals.

**Organ:** A group of tissues working together e.g. heart.

**Ovulation:** This is the release of an egg from an ovary.

**Phloem:** This is a plant transport tissue. It transports food from where it is made to other parts of the plant.

**Photosynthesis:** This is the process in which green plants make food.

**Phototropism:** The growth of a plant in response to light.

**Placenta:** The structure that binds the developing baby to the wall of the womb. It allows nutrients and waste to be exchanged.

**Pollination:** This is the transfer of pollen from the anther of the stamen to the stigma of the carpel.

**Pollution:** This is where things such as oil, sewage, slurry, sulphur dioxide, nitrogen oxides and litter damage the environment.

**Pooter:** A piece of equipment used to collect small animals.

**Producer:** An organism that can make its own food.

**Respiration:** This is a chemical process where energy is released from food.

**Retina:** The light sensitive layer at the back of the eye.

**Sensory nerve:** A nerve that carries messages to the brain and spinal cord.

**Soda lime:** This is used to absorb carbon dioxide.

**Stamen:** The male part of the flowering plant. It is made up of the anther and filament.

**Stomata:** These are pores (openings) in the leaves of a plant, which allows gases to diffuse.

**Synovial fluid:** A lubricating fluid found at a joint. It helps reduce friction.

**System:** A group of organs working together e.g. digestive system.

**Tendons:** Fibres that attach muscle to bone.

**Tissue:** A group of similar cells e.g. muscle.

**Transpiration:** This is the loss of water vapour from the surface of a plant. It is highest when there is a gentle breeze, sun and low humidity.

**Trophic level:** The position an organism occupies in a food chain.

**Tropism:** A growth response to a stimulus.

**Tullgren funnel:** A piece of equipment used to extract small animals from leaf litter or soil.

**Xylem:** This is a plant transport tissue. It transports water and minerals from the roots to other parts of the plant.

**Zygote:** The cell, which results from the fusion of a male and female gamete.